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Solution: Since reaction is balanced, number of atoms in both sides

must be equal.  $C_xH_y(OH)_z + 5O_2 \rightarrow 4CO_2 + 5H_2O$ . In products

we have; 13 O atoms, there must be 13 O atoms in reactants. But we

have 10, so "z" must be 3. In products we have 4 C atoms, to

balance it there must be 4 in reactants, so "x" is 4.

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pdf. published by Brooks/Cole 2012. According to authors; The authors of this book have many years of experience teaching general chemistry and other areas of chemistry at the college level.

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Equal amounts of two acids—HCl and HCO<sub>2</sub>H (formic acid) ...  
In addition, a recent tenure as a mentor of the U.S. Chemistry Olympiad Team, Professor Kotz has received numerous honors, including a State University of New York Chancellor's Award (1979), a National Catalyst Award for Excellence in Teaching (1992), the Estee Lectureship in Chemical Education at the University of South Dakota (1998), the Visiting Scientist Award from the Western Connecticut Section of the American Chemical Society (1999), and the first annual Distinguished Education Award ...

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