

## Chemquest 22 Covalent Bonding Answer Key

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Covalent vs. Ionic bondsChemical Bonding - Ionic vs. Covalent Bonds  
Ionic vs. Molecular

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? Covalent Bond and its Types ? Lewis Structure ? Chemistry for Class 11 in HINDIHow to Draw Covalent Bonding Molecules **Chemquest 22 Covalent Bonding Answer**

A covalent bond is the type of bond between two nonmetals. Covalent bonds are formed by neutral atoms that share electrons rather than by charged ions. When a compound is formed by sharing electrons, the compound is called a molecule or molecular compound. It is important to note that ionic compounds are not called molecules.

### Information : Terminology

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The number of electrons that should appear in the bonding picture for CO<sub>3</sub> is 22. The number of electrons that appear in the picture for CO<sub>3</sub><sup>2-</sup> is 24. Offer an explanation for why CO<sub>3</sub><sup>2-</sup> has 24 electrons instead of 22. (Where did the extra two electrons come from?)

### ChemQuest 23 : Drawing Covalent Compounds

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### Chemquest 23 Lewis Structures Answer Key

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