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Copyright © 2012 Nelson Education Ltd. Chapter 7: Chemical Equilibrium 7.5-2 Statement: The value of Q is 0.867. Q is approximately equal to K, so the reaction is at equilibrium. 3. (a) The equilibrium constant equation for the reaction represented by ! 2 Ag 20(s) #!! 4 Ag(s) + 0 2(g) is K=[0 2 (g)], because the other substances are solids.

Section 7.5: Quantitative Changes in Equilibrium Systems -

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The video below refers to Dynamic Equilibrium (under the section Equilibrium Systems) 7.1 p. 420 in the Nelson Textbook The video below is related to 7.2 Equilibrium Law and the Equilibrium Constant (p.429 in the Textbook)

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